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REVIEW ARTICLE

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## REVIEW ON VARIOUS ANALYTICAL AND BIOANALYTICAL METHODOLOGIES FOR FLUCONAZOLE

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### ABSTRACT

Fluconazole is an azole antifungal drug, used to treat serious fungal or yeast infections, such as those affecting the vagina, mouth or throat, lungs, bladder, or blood. It comes in injectable, topical and oral dosage forms. The development and validation of analytical methods are ongoing, interrelated processes carried out during the drug development process. Establishing the suggested analytical method's accuracy, specificity, precision, and robustness for the pharmaceutical industry's analysis of a drug moiety is the main goal of its development and validation. The creation of a methodology is necessary for the pharmaceutical formulation's drug discovery, development, and assessment. Validation is required to demonstrate that an analytical method is appropriate and suitable for the intended use, which is frequently a crucial prerequisite for analytical purposes. This article focuses on the development of different analytical method development techniques and validation of fluconazole.

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## INTRODUCTION

The triazole class of antifungals includes fluconazole,  $\alpha$ -(2,4-difluorophenyl)- $\alpha$ -(1H-1,2,4-triazol-1-yl)-1H-1,2,4-triazol-1-ethanol. It is recommended for oropharyngeal candidiasis, esophageal, vaginal, and deep infections and exhibits activity against species of *Candida*. Ergosterol, a steroid that is specific to fungal cell membranes, is selectively inhibited by fluconazole. Developed in the 1980s, fluconazole is a triazole that exhibits activity against species of *Candida* and is available in oral tablet, capsule and suspension forms, topical gel, lotion and dusting powder, and injection for intravenous infusion (1).

**Physicochemical Properties:** Among the triazole class, fluconazole is a member of propan-2-ol that has a 2,4-difluorophenyl group at position 2 and 1H-1,2,4-triazol-1-yl groups at positions 1 and 3. It has a solid, white, crystalline powder appearance. It has molecular weight of 306.27 g/mol. The melting point is between 138-140 °C. The pKa value is 1.76. Although it dissolves poorly in water, it dissolves in organic solvents including methanol, propylene glycol, chloroform, and others (2).

**Pharmacokinetic:** Fluconazole has an oral bioavailability of over 90%, according to measurements. Fluconazole's protein binding is thought to be between 11-12%. When given over an extended period of time, fluconazole's significant penetration in a variety of bodily

fluids makes it an excellent treatment for systemic fungal infections. The liver metabolizes fluconazole minimally. Fluconazole is a CYP2C9, CYP3A4, and CYP2C19 inhibitor. About 80% of the administered dose of fluconazole is found in the urine as unmodified medication in healthy volunteers, where the drug is excreted through the kidneys. After oral treatment, the terminal elimination half-life in plasma is roughly 30 hours (range: 20–50 hours). (3)

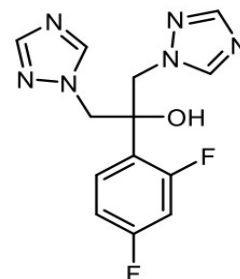


Figure 1. Chemical Structure of Fluconazole

**Pharmacodynamic:** Fluconazole exhibits fungistatic activity, against various microorganisms: *Candida albicans*, *Candida glabrata*, *Candida Parapsilosis*, *Candida tropicalis*, and *Cryptococcus neoformans*. In order to treat fungal infections and their symptoms, steroidal inhibition in fungal cells is used to disrupt cell adhesion, proliferation, and synthesis of the cell wall. In normal and

immunocompromised animal models, fluconazole has also demonstrated fungistatic action against intracranial and systemic fungal infections.(4)

**Mechanism of action:** The fungal cytochrome P450-dependent enzyme lanosterol 14- $\alpha$ -demethylase is highly selectively inhibited by fluconazole. Normally, this enzyme converts lanosterol to ergosterol, which is required for the formation of fungal cell walls. One iron atom in the heme group of lanosterol 14- $\alpha$ -demethylase forms a bond with the free nitrogen atom on the azole ring of fluconazole. This stops the process of ergosterol manufacture by preventing oxygen activation, which in turn stops lanosterol from being demethylated. It is then discovered that methylated sterols build up in the fungal cellular membrane, which inhibits the growth of the fungus. The structure and functionality of the fungal cell plasma membrane are negatively impacted by these accumulating sterols (5).

**Need of analytical method development and validation:** During the drug discovery, release to market, and development processes that lead to marketing approval, the development of sound analytical method or methods is crucial. In order to support routine, in-process, and stability analysis, methods for determining the selectivity, specificity, limit of detection, limit of quantitation, linearity, range accuracy, precision, recovery solution stability, ruggedness, and robustness of various analytical methods are being developed. These methods are intended for use in finished product or process tests, as well as in the preparation of drug products (6).

UV-visible spectroscopy is used to determine the absorbance spectrum of a chemical in solid or solution form. (7)

**Analytical Method Development by HPLC:** The most popular analytical method is High Performance Liquid Chromatography (HPLC). HPLC is a separation method that primarily consists of stationary and mobile phases with opposite polarities and uses high pressure pumps for separation. The interaction between the stationary and mobile phases allows for separation. Variables that are crucial to the creation of a method include the pH of the mobile phase, various types of buffer, column temperature, sample diluents, detection wavelength, and many more (15).

**Analytical Method Development by HPTLC:** Based on the full potential of thin layer chromatography, high performance thin layer chromatography (HPTLC) is an advanced instrumental technique. It is a potent analytical instrument for chromatographic information of complex mixtures of inorganic, organic, and biomolecules because of its benefits, which include automation, scanning, comprehensive optimization, the selective detection principle, minimal sample preparation, hyphenation, etc. HPTLC has great promise for estimating partitioning qualities to enable investigations on environmental destiny, combinatorial chemistry, and health effects. (21)

**Analytical method development using bioanalytical methods:** In the process of discovering and developing new drugs, bioanalysis is

**Table 1. Analytical method development using UV-spectrophotometer**

S. No.	Sample / Dosage form	Method / Instrument model	Solvent / Solution	Wavelength (nm)	References
1.	Tablet	UV-Visible Spectrophotometer (UV-1700 SHIMADZU)	Phosphate buffer pH 7.4	260	(8)
2.	Bulk	UV-visible spectrophotometer	0.1N HCl	239	(9)
3.	Bulk	UV-Visible Spectrophotometer, (Model UV1800)	Water	260.8	(10)
4.	Tablets	UV-visible spectrophotometer 1700 Ultra	Ethanol	252	(11)
5.	Tablets	Spectrophotometer UV-Vis Cary 100 Conc (Varian Analytical Instruments, Palo Alto, USA)	Methanol	261	(12)
6.	Tablets	Shimadzu UV 1700 double beam spectrophotometer	Phosphate buffer pH 6.8	260.57 & 264.23	(13)
7.	Capsules and IV Solutions	Shimadzu UV-160A recording double beam UV-visible spectrophotometer	Distilled water	261.6 & 274	(14)

**Table 2. Analytical method development using HPLC method**

S.No.	Sample	Stationary phase/column	Mobile phase	Wavelength (nm)	Flow rate (ml/min)	RT (min)	Reference
1.	Micro-emulsions & liquid crystals	C18-RP column (250 × 4.6 mm)	Acetonitrile: water (50:50, v/v)	210	1.0	6.3	(16)
2.	Tablets and bulk	C18 column (4.6 × 250 mm, 5 $\mu$ m)	Acetonitrile and phosphate buffer pH 4.8 (50:50 % v/v)	210	1.0	5.25	(17)
3.	Capsules, IV infusions and bulk	C18 (10 mm, 25 cm, 4.6 mm)	Acetonitrile and 25 mMtris-hydroxymethyl aminomethane in phosphate butter pH 7 (55:45 v/v)	260	1.5	2.474	(18)
4.	Tablets and capsules	C-18 (150 mm x 4.6 mm, 5 $\mu$ m)	Water and acetonitrile (65:35 v/v)	260	1.0	2.47	(19)
5.	Tablets	C18 column (4.6 × 250 mm, 5 $\mu$ m)	0.05 M potassium dihydrogen phosphate buffer (pH 3.25 and acetonitrile (82:18, v/v).	210	1.5	5.38	(20)

**Table 3. Analytical Method Development Using HPTLC Method**

S.No.	Sample	Stationary Phase/ Column	Mobile phase	Wavelength (nm)	Reference
1.	Tablets and bulk	Silica gel 60 F254	Butanol: water: acetic acid (8:2:1 v/v)	254	(22)
2.	Tablets or capsules and bulk	Silica gel 60 GF254	Toluene: chloroform: methanol 1.2:3.0:0.4 (v/v)	210	(23)
3.	Tablets	Silica gel 60F254	Toluene: Ethyl acetate: Chloroform: Methanol (1.2:3:2:0.8 v/v)	205	(24)

**Analytical Method Development by UV-Spectrophotometry:** The pharmaceutical sector has been using UV spectroscopy as an important and efficient analytical tool. This pharmaceutical analysis includes all of the procedures needed to ascertain the identity, strength, quality, or purity of the substances. The study methodology uses colorless materials to test their absorption of monochromatic light in the 200–400 nm band of the near-ultraviolet spectrum.

crucial. The examination of analytes in biological samples is known as bioanalysis, and it entails a number of procedures, including sample collection, analysis, and data reporting. Samples from clinical or preclinical investigations are first collected, and they are subsequently sent to a lab for analysis. The second step is, sample cleanup or preparation. A steady and robust sample preparation technique should be used to get accurate results. The final step is the analysis and detection of the material.(25)

Table 4. Bioanalytical Method Development

S.No.	Method	Sample	Stationary phase/ column	Mobile phase	Wavelength (nm)	Flow rate (ml/min)	RT (min)	Reference
1.	LC-MS/MS	Human Plasma	C18 (100 x 2.1mm, 3.5 $\mu$ m)	0.1% Formic acid water and acetonitrile with 0.1% formic acid (70:30 v/v)	-	0.2	-	(26)
2.	HPLC	Biological Skin Matrices	RP 18 column (5 $\mu$ m, 250 x 4 mm)	Methanol and 0.025 mol/L phosphate buffer (45:55, v/v), pH 7.0	260	1.0	5.1	(27)
3.	HPLC - UV	Human plasma and urine	ODS 100 A column (250 x 4.6 mm, 5 $\mu$ m)	0.15M SDS, 0.3% TEA and 12% n-propanol in 0.02M orthophosphoric acid at pH 5.5	210	1.0	3.89	(28)
4.	HPLC	Human plasma	C18 column (150 x 4.6 mm)	10 mM acetate buffer at pH 5.0 and methanol (65:35 v/v)	210	1.0	7.00	(29)
5.	HPLC	GFS, VFS, topical simulated media, PBS	C18 column (250 mm x 4.6 mm x 5 $\mu$ m)	Water and acetonitrile, pH 5.2 (80:20, v/v)	260	2.5	8.05 8.02 7.98 7.97	(30)
6.	HPLC	Human Whole Blood	C18 (5 $\mu$ m, 4.6 x 250 mm)	Acetonitrile and water (36:64, v/v)	210	0.8	4.7	(31)
7.	HPLC-UV	Human plasma	C18 (250 x 4.6 mm, 5 $\mu$ m)	pH 5.0 with glacial acetic acid buffer-acetonitrile (80:20, v/v)	260	1.2	8.3	(32)
8.	HPLC	Human Urine	C18 (150 mm x 4.6 mm, 5 $\mu$ m)	Methanol: water (70:30, v/v)	254	0.85	-	(33)
9.	HPLC	Intracellular concentrations	C18 (250 x 2 mm, 3 $\mu$ m)	0.01 mol/L ammonium acetate in water (pH5) and acetonitrile	210	0.3	6.8	(34)
10.	HPLC	Human plasma	C18 (250 mm x 4.6 mm, 5 $\mu$ m)	Acetonitrile and 10 mM sodium phosphate buffer (30:70, v/v)	210	1.0	4.6	(35)
11.	LC-MS	Human Plasma	50 x 4.6 mm, 4 $\mu$ m	Methanol, 0.1% v/v formic acid	-	1.0	-	(36)

## CONCLUSION

Drug development requires the development, validation, and transfer of analytical methods. For a biotech company, method development is essential for patient safety, regulatory, and commercial reasons. Approval of clinical trials and marketing authorizations are unattainable without the development and validation of high-quality methods. The many analytical techniques for estimating fluconazole in bulk and in its varied dosage forms were covered in this article. These techniques could be useful for fluconazole analysis in a variety of study fields.

**Conflict of interests:** The author reported no conflict of interests.

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